

Investigation of the Microwave Spectrum of Cyclopropyl Isocyanate. An Example for the Failure of Centrifugal Distortion Theory

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The vibrational ground state microwave spectrum of cyclopropyl isocyanate was investigated in the region from 8.4 to 40 GHz by microwave Fourier transform (MWFT) spectroscopy. The quadrupole hyperfine structure was resolved and assigned. The quadrupole coupling constants are given. With respect to the data given in the literature up to now, this work led to a more profound description of the pure rotational spectrum. Furthermore, some interesting and surprising results concerning the effect of centrifugal distortion are presented. Compared to hitherto existing investigations, these results indicate a more complicated conformational behaviour of cyclopropyl isocyanate.

Introduction

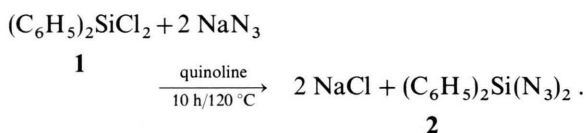
The first investigations of cyclopropyl isocyanate in 1988 by microwave, infrared and Raman spectroscopy were carried out by Durig, Berry and Wurrey [1]. They presented data about the conformational stability, structure, dipole moment, as well as the vibrational assignment of cyclopropyl isocyanate. From known structural parameters of cyclopropyl isothiocyanate [2], Durig et al. evaluated sets of rotational constants for various assumed conformers and compared them with the rotational constants obtained from the assignment of the microwave spectrum in the region from 18.5 to 40 GHz. The *cis* and *trans* conformation corresponded best with the observed values.

The results of our reinvestigation, and in particular the failure of the centrifugal distortion analysis, indicate a more complicated conformational behaviour of cyclopropyl isocyanate. Especially, there are some doubts about the existence of a *cis* conformation. Thus, the expression “*cis*” is set in quotation marks in this paper.

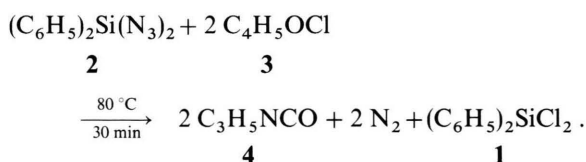
Another important item concerning this investigation is the resolution and assignment of the quadrupole hyperfine structure of the spectrum of cyclopropyl isocyanate since no quadrupole coupling constants were determined up to now.

Experimental Methods

Cyclopropyl isocyanate was prepared according to a method of Kricheldorf and Regel [3, 4] by a two-step procedure. In the first step dichloro diphenyl silane **1** (Aldrich) reacts with sodium azide in quinoline to yield diphenyl diazido silane **2**:



The solvent was dried with calcium hydride for about two hours to remove traces of water. Cyclopropyl isocyanate **4** is formed by a reaction according to Curtius of cyclopropyl carbonyl chloride **3** (Aldrich) with the intermediate product **2**:



The product **4** was purified under vacuum in a fraction column. The purity was checked by NMR and gas-chromatography. No impurities were found with these methods. Cyclopropyl isocyanate is a colourless liquid (b.p.: 78–79 °C). The substance can immediately be identified by its irritating effect and provocation of laceration. The compound is sensitive to traces of water and was stored in a sample tube at –178 °C (liq. nitrogen).

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The investigations of cyclopropyl isocyanate were carried out with our standard microwave Fourier transform spectrometers combined with double-resonance techniques in the region from 8.4 to 40 GHz [5–9]. Sample pressures in the waveguide spectrometers were often <0.3 mTorr (<0.04 Pa) and temperatures were about -50°C .

The transition frequencies of the multiplet components were evaluated by a least squares fit of the time domain signal [10] to avoid overlapping effects.

In addition, some results were obtained with a beam spectrometer described in [11]. For the measurements we used argon containing ca. 1% cyclopropyl isocyanate at a stagnation pressure of 380 Torr ($5 \cdot 10^4$ Pa).

Results and Analysis

Figure 1 shows the conformers trans and cis cyclopropyl isocyanate in their principal axes system. The notation cis/trans characterizes the relative orientation of the cyclopropyl frame and the $\text{N}=\text{C}=\text{O}$ group with respect to the $\text{N}-\text{C}$ bond. The trans conformer possesses an a , c -, the cis conformer an a , b -plane of symmetry. Furthermore, the polar NCO group lies approximately in the direction of the a -axis. Thus, a -type transitions should be the most intense ones in the microwave spectrum. The remaining components μ_c (trans) and μ_b ("cis") are expected to be smaller, whereas the μ_b (trans) and μ_c ("cis") components are zero because of symmetry properties. The microwave investigation of Durig et al. leads to the dipole moments: $\mu_a = 2.56(2)$ D, $\mu_c = 0.71(3)$ D, $\mu_b = 0$ D for the trans and $\mu_a = 2.720(4)$ D, $\mu_b = 0.17(1)$ D, $\mu_c = 0$ D for the "cis" isomer.

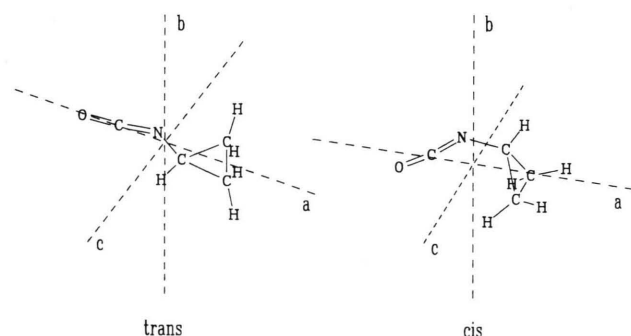


Fig. 1. Schematic diagram of trans and cis cyclopropyl isocyanate in their principal axes system. The molecular structure is taken from the data in [1].

For the reason of intensity we tried to resolve and assign the quadrupole hyperfine structure of the a -type spectrum. We started with the trans conformer and obtained the quadrupole coupling constants $\chi^+ = (\chi_{bb} + \chi_{cc})$ by an iterative measurement and fit procedure of R-branch series with low J . $\chi^- = (\chi_{bb} - \chi_{cc})$ was not accessible, the observed transitions being nearly independent of this constant. Thus, we attempted to measure c -type transitions. Because of the large uncertainty of the rotational constant A [1], we used double-resonance techniques with the pump/signal combination in the V/Ku band. We found the Q-branch series $J_{1J} - J_{0J}$ with $J = 3$ to $J = 12$, which enabled us to gradually evaluate the A and χ^- constants. Subsequently, the value of χ^- was refined by measurements of the $J' - J = 3 - 2$ transitions with a beam spectrometer using its high capability for resolution and sensitivity.

The investigation of the "cis" conformer was more difficult. At the beginning, no consistent assignment of the quadrupole hyperfine structure, even for the a -type spectrum, was possible. The A constant was not well determined by Durig et al., and there were no suitable double-resonance combinations which could simplify the search for the b -type transitions. After some unsuccessful attempts with conventional continuous-wave Stark modulation techniques and continuous-wave double-resonance methods in the Ku/K band (pump/signal) we refrained from measurements of b -type transitions and evaluation of the A constant. An explanation of this unsuccessful experimental result may be the lower theoretical line strength in comparison with a -type transitions (factor: ≈ 260).

Moreover, it was not possible to observe any transitions of "cis" cyclopropyl isocyanate in the beam spectrometer. Thus, the trans must be more stable than the "cis" conformation. This corresponds with the data of Durig et al. The impossibility of the observation of "cis"-transitions in a beam spectrometer leads to another fact: during the short time of cooling down in the molecular beam, the majority of the "cis-molecules" is apparently transformed into the trans conformation. Thus, the energy barrier should be low between the two conformational states.

On account of the above mentioned difficulties of the "cis" conformer we measured a -type R-branch J -series ($J_{K_+ K_+} - (J-1)_{K_+ K_+}$) for the "cis" ($J = 5, 6, 7, 8, 9$) and trans ($J = 8, 9, 10, 11$) isomer in the K and V band. We started for each J value with the $K_- = 0$ transition, then searched for the $K_- = 1$ lines and so

on. The transitions were assigned by means of their quadrupole hyperfine splittings. This is corroborated by Fig. 2 representing the $K_- = 3$ transitions for $J = 7, 8, 9$. The quadrupole hyperfine structure is not resolvable for $K_- = 0, 1$ transitions. In general, the splitting of the $K_- = 2$ transitions can be observed, but they are very small. Therefore, the extrapolation to larger splittings could lead to uncertainties and incorrect assignments. For this reason the typical pattern of the $K_- = 3$ lines (Fig. 2) is of particular importance for this investigation. The $K_- = 3$ lines are characterized by a small K-doublet splitting combined with a sufficient quadrupole hyperfine splitting ensuring a correct extrapolation. A comparison of Figs. 2a, 2b, and 2c indicates that according to theory the asymmetry splitting increases and the quadrupole splitting decreases with increasing J value. For all measured transitions a consistent assignment with respect to the fit of the quadrupole coupling constants could be evaluated in combination with double-resonance techniques.

Table 1 contains the evaluated values of the quadrupole coupling constants for trans and "cis" cyclopropyl isocyanate.

Carrying out the measurement procedure by following increasing K_- as described above, we obtained a very surprising result. The frequencies of the $K_- = 0, 1, 2$ transitions correspond (within the standard error) with the prediction of the spectrum including centrifugal distortion effects. The frequencies of the lines with

Table 1. Quadrupole coupling constants of trans and "cis" cyclopropyl isocyanate. Numbers in parentheses represent single standard errors in units of the last quoted digit.

<i>trans</i> cyclopropyl isocyanate	Correlation matrix	
	χ^+	χ^-
$\chi^+ = -2.6306(26)$ MHz	1.00	
$\chi^- = -0.1372(36)$ MHz	0.07	1.00
Standard deviation of the fit: 3 kHz	$\chi_{aa} = 2.6306(26)$ $\chi_{bb} = -1.3839(31)$ $\chi_{cc} = -1.2467(31)$	
<i>"cis"</i> cyclopropyl isocyanate	Correlation matrix	
	χ^+	χ^-
$\chi^+ = -2.5647(49)$ MHz	1.00	
$\chi^- = 0.440(10)$ MHz	0.00	1.00
Standard deviation of the fit: 5 kHz	$\chi_{aa} = 2.5647(49)$ $\chi_{bb} = -1.0624(75)$ $\chi_{cc} = -1.5022(75)$	

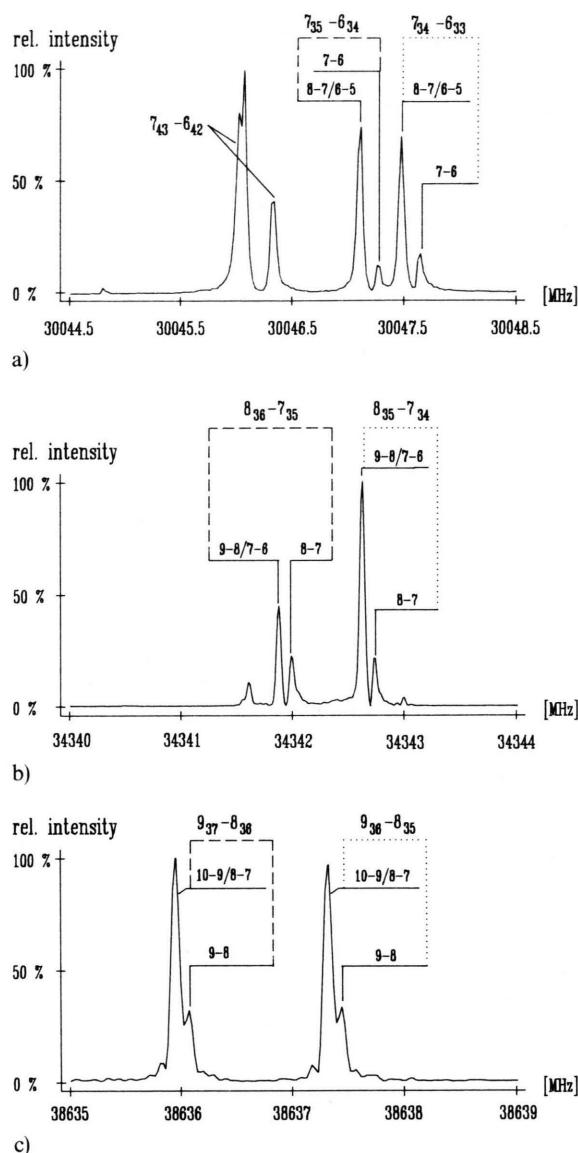


Fig. 2 a–c. The $K_- = 3$ transition for $J = 7, 8, 9$. The $K_- = 3$ transitions are characterized by a small K-doublet splitting in combination with a quadrupole hyperfine splitting. This typical pattern was important for the confirmation of the assignment. a) Transitions $7_{35}-6_{34}$ and $7_{34}-6_{33}$ (and $7_{43}-6_{42}$): 0.2 mTorr, -50°C , polarisation frequency: 30 048.5 MHz, sample interval: 10 ns, 1024 data points supplemented by 3072 zeros before Fourier transformation, 76 800 k experiment cycles (1 k = 1024). b) Transitions $8_{36}-7_{35}$ and $8_{35}-7_{34}$: 0.2 mTorr, -50°C , polarisation frequency: 34 343 MHz, sample interval: 10 ns, 1024 data points supplemented by 3072 zeros before Fourier transformation, 45 440 k experiment cycles. c) Transitions $9_{37}-8_{36}$ and $9_{36}-8_{35}$: 0.2 mTorr, -50°C , polarisation frequency: 38 637 MHz, sample interval: 10 ns, 1024 data points supplemented by 3072 zeros before Fourier transformation, 16 600 k experiment cycles.

$K_- > 2$ showed considerable deviations from the prediction depending on the value of K_- . In general, the deviations increase with K_- , but there are also some differences in the K_- -dependence of the deviations for the trans and "cis" conformer. Figures 3a and 3b present the dependence of the deviation $\Delta\nu$ (difference between the calculated central frequency and the ob-

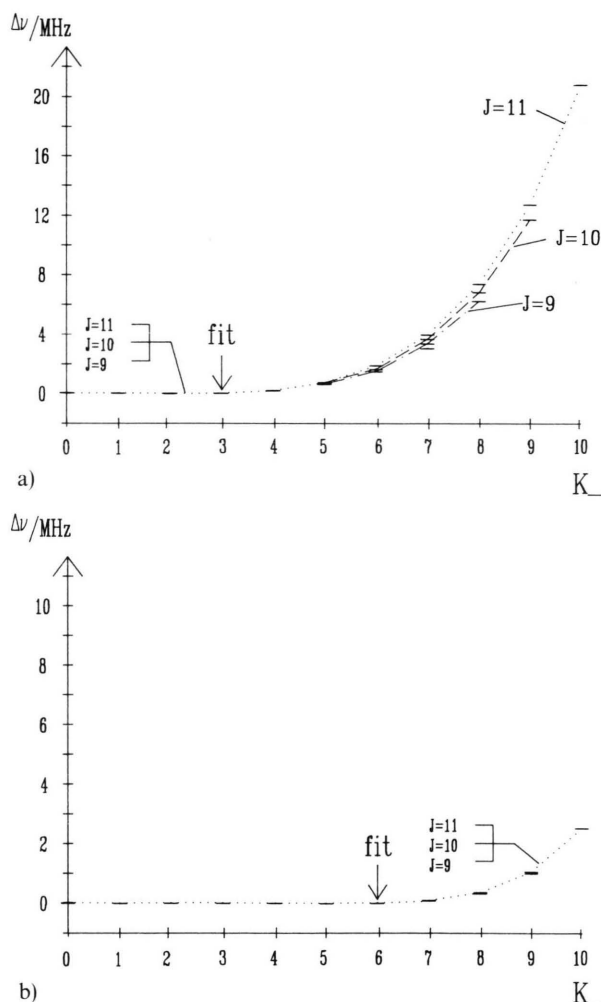


Fig. 3a and b. Deviations $\Delta\nu$ from a fit using Van Eijck's centrifugal distortion Hamiltonian for trans cyclopropyl isocyanate. The deviations strongly depend on the angular momentum projection quantum number K_- (in this figure only presented for $J=9, 10, 11$). Table 3 contains the evaluated spectroscopic constants for the different fits. a) Deviations from a fourth order centrifugal distortion fit. Transitions with $K_- > 3$ are weighted with the factor 1/1000 in the fit. This is indicated by the arrow \downarrow . b) Deviations from a sixth order centrifugal distortion fit. Transitions with $K_- > 6$ are weighted with the factor 1/1000 in the fit. This is indicated by the arrow \downarrow . In comparison to a), the axis concerning $\Delta\nu$ is spread by the factor 2.

served value) on the quantum number K_- for different J values for the trans conformer. Figure 3a is based on a fit using Van Eijck's fourth order centrifugal distortion Hamiltonian, Fig. 3b implies terms of Van Eijck's sixth order Hamiltonian. In Fig. 3a transitions with $K_- > 3$ are weighted with the factor 1/1000 in the fit (which actually means: excluded from the fit) to get an acceptable standard deviation for the fit. This is indicated by the arrow \downarrow in the picture. Figure 3a shows that the deviation $\Delta\nu$ gets large with increasing K_- (e.g. for $K_- = 10$: $\Delta\nu = 22$ MHz) but in a way indicating a functional continuity. $\Delta\nu$ is extremely decreased by consideration of the sixth order centrifugal distortion Hamiltonian (the $\Delta\nu$ -axis is spread by the factor 2 in Figure 3b). In addition, it is possible to consider a greater number of transitions in the fit. Transitions with $K_- > 6$ are still weighted with the factor 1/1000. Hopefully, an eighth order centrifugal distortion Hamiltonian will eliminate the remaining deviations.

Table 2 contains all measured transition frequencies of the trans conformer including the quadrupole hyperfine structure, the (hypothetical) central frequency ν_0 and the calculated central frequencies resulting from a fourth (${}^4\nu_0$) and sixth (${}^6\nu_0$) order centrifugal distortion fit. The evaluated sets of the rotational as well as Van Eijck's centrifugal distortion constants for the different fit procedures (according to Figs. 3a and 3b) are given in Table 3.

The deviations from a Van Eijck fourth order centrifugal distortion fit for the "cis" isomer is illustrated by Figure 4. The fit includes all measured transition frequencies for K_- values up to $K_- = 2$. The relatively

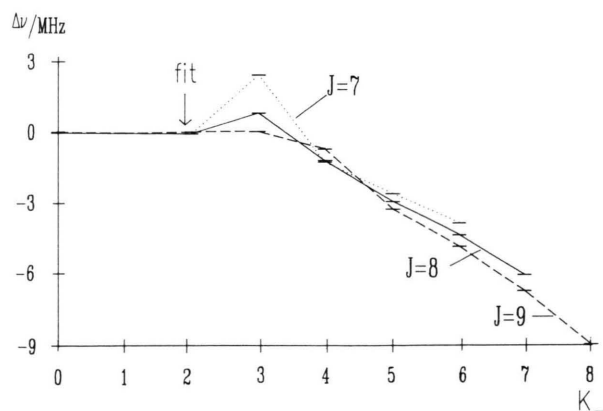


Fig. 4. Deviations $\Delta\nu$ from a fourth order centrifugal distortion fit in dependence on K_- for "cis" cyclopropyl isocyanate. Transitions with $K_- > 2$ are weighted by the factor 1/100. This is indicated by the arrow \downarrow .

Table 2 (continued)

J	K ₋	K ₊	J'K ₋ 'K ₊ '	F'-F	ν_{obs}	δ_{HFS}	$\Delta(\delta_{\text{HFS}})$	ν_0 $\frac{4\nu_0}{6\Delta\nu_0}$	$\frac{4\Delta\nu_0}{6\Delta\nu_0}$	J	K ₋	K ₊	J'K ₋ 'K ₊ '	F'-F	ν_{obs}	δ_{HFS}	$\Delta(\delta_{\text{HFS}})$	ν_0 $\frac{4\nu_0}{6\Delta\nu_0}$	$\frac{4\Delta\nu_0}{6\Delta\nu_0}$
					[MHz]	[MHz]	[kHz]	[MHz]	[MHz]						[MHz]	[MHz]	[kHz]	[MHz]	[MHz]
5	2	3	4 2 2	6-5 4-3 5-4	17504.050 << 17504.240			17504.106 17504.102 17504.111	-0.004 0.005	11	3	8	10 3 7	12-11 10-9 11-10	38500.750 38500.750 38500.790			38500.763 38500.764 38500.768	0.001 0.005
5	2	4	4 2 3	6-5 4-3 5-4	17499.480 << 17499.671			17499.536 17499.531 17499.539	-0.005 0.003	11	3	9	10 3 8	12-11 10-9 11-10	38500.138 38500.138 38500.180			38500.153 38500.153 38500.156	-0.000 0.003
6	2	4	5 2 3	7-6 5-4 6-5	21006.951 << 21007.069			21006.993 21006.979 21006.991	-0.014 -0.002	5	4	2	4 4 1	6-5 5-4 4-3	17493.959 17494.663 17493.752			17494.131 17494.267 17494.131	0.136 -0.000
6	2	5	5 2 4	7-6 5-4 6-5	20998.958 << 20999.067			20998.992 20998.981 20998.990	-0.011 -0.002	8	4	4	7 4 3	9-8 8-7 7-6	27991.196 27991.374 27991.166			27991.248 27991.432 27991.244	0.184 -0.004
8	2	6	7 2 5		n. r.			28016.292 28016.274 28016.293	-0.018 0.001	9	4	5	8 4 4	9-8 10-9 8-7	31490.550 31490.422 31490.415			31490.463 ^{**)} 31490.658 31490.460	0.195 -0.003
8	2	7	7 2 6		n. r.			27997.106 27997.093 27997.105	-0.013 -0.001	10	4	6	9 4 5	11-10 9-8 10-9	34989.755 34989.755 34989.853			34989.789 34989.988 34989.787	0.199 -0.002
9	2	7	8 2 6		n. r.			31523.071 31523.056 31523.079	-0.015 0.008	11	4	7	10 4 6	12-11 10-9 11-10	38489.209 38489.209 38489.278			38489.230 38489.435 38489.235	0.205 0.005
9	2	8	8 2 7		n. r.			31495.681 31495.672 31495.686	-0.009 0.005	8	5	3	7 5 2	8-7 9-8 7-6	27982.994 27982.718 27982.678			27982.798 ^{**)} 27983.413 27983.791	0.615 -0.007
10	2	8	9 2 7		n. r.			35031.521 35031.495 35031.524	-0.026 0.003	9	5	4	8 5 3	9-8 10-9 8-7	31480.964 31480.768 31480.744			31480.827 ^{**)} 31481.489 31480.814	0.662 -0.013
10	2	9	9 2 8		n. r.			34993.891 34993.878 34993.894	-0.013 0.003	10	5	5	9 5 4	10-9 11-10 9-8	34979.006 34978.864 34978.846			34978.907 ^{**)} 34979.619 34978.899	0.712 -0.008
11	2	10	10 2 9		n. r.			38491.685 38491.669 38491.687	-0.016 0.002	11	5	6	10 5 5	12-11 10-9 11-10	38477.024 38477.024 38477.137			38477.062 38477.807 38477.052	0.745 -0.010
5	3	2	4 3 1	6-5 5-4 4-3	17498.080 17498.487 17497.980			17498.189 17498.211 17498.192	0.022 0.003	7	6	1	6 6 0	8-7 7-6 6-5	24475.666 24476.260 24475.551			24475.838 24477.159 24475.841	1.321 0.003
8	3	5	7 3 4	9-8 7-6 8-7	27998.435 << 27998.542			27998.472 27998.491 27998.474	0.019 0.002	8	6	2	7 6 1	9-8 8-7 7-6	27972.357 27972.752 27972.292			27972.470 27973.953 27972.476	1.483 0.006
8	3	6	7 3 5	9-8 7-6 8-7	27998.318 << 27998.428			27998.357 27998.371 27998.354	0.014 -0.003	9	6	3	8 6 2	9-8 10-9 8-7	31469.337 31469.060 31469.021			31469.140 ^{**)} 31470.771 31469.145	1.631 0.005
9	3	6	8 3 5	10-9 8-7 9-8	31498.943 << 31499.014			31498.965 31498.983 31498.970	0.018 0.005	11	6	5	10 6 4	11-10 12-11 10-9	38462.708 38462.550 38462.535			38462.600 ^{**)} 38464.486 38462.604	1.886 0.004
9	3	7	8 3 6	10-9 8-7 9-8	31498.725 << 31498.796			31498.746 31498.763 31498.750	0.017 0.040	8	7	1	7 7 0	9-8 8-7 7-6	27959.742 27960.277 27959.646			27959.893 27962.926 27959.972	3.033 0.079
10	3	7	9 3 6	11-10 9-8 10-9	34999.702 < 34999.756			34999.720 34999.728 34999.723	0.008 0.003	9	7	2	8 7 1	10-9 9-8 8-7	31454.853 31455.230 31454.798			31454.961 31458.319 31455.048	3.358 0.087
10	3	8	9 3 7	11-10 9-8 10-9	34999.325 << 34999.379			34999.343 34999.352 34999.346	0.009 0.003	10	7	4	9 7 3	11-10 10-9 9-8	34949.983 34950.258 34949.949			34950.063 34953.723 34950.151	3.660 0.088

Table 2 (continued)

J	K ₋	K ₊	J'K' ₋ 'K ₊ '	F'-F	ν_{obs} [MHz]	δ_{HFS} [MHz]	$\Delta(\delta_{\text{HFS}})$ [kHz]	ν_0 [MHz]	$\delta \nu_0$ [MHz]
11	7	4	10 7 3	12-11	38445.117			38445.182	
				11-10	38445.335	-0.218	-8	38449.138	3.956
				10-9	18445.090	0.027	5	38445.285	0.103
9	8	1	8 8 0	10-9	31437.716			31437.859	
				9-8	31438.210	-0.494	-1	31444.063	0.6204
				8-8	31437.639	0.077	1	31438.191	0.332
10	8	2	9 8 1	11-10	34930.946			34931.050	
				10-9	34931.305	-0.359	2	34937.847	6.797
				9-8	34930.892	0.054	6	34931.405	0.355
11	8	3	10 8 2	12-11	38424.196			38424.273	
				11-10	38424.464	-0.268	5	38431.630	7.357
				10-9	38424.158	0.038	7	38424.646	0.373
10	9	1	9 9 0	11-10	34908.105			34908.237	
				10-9	34908.560	-0.455	1	34919.940	11.703
				9-8	34908.041	0.064	0	34909.237	1.000
11	9	2	10 9 1	12-11	38399.090			38399.191	
				11-10	38399.433	-0.343	2	38411.902	12.711
				10-9	38399.047	0.043	0	38400.259	1.068
11	10	1	10 10 0	12-11	38369.023			38369.142	
				11-10	38369.448	-0.425	0	38389.924	20.782
				10-9	38368.968	0.055	2	38371.676	2.534

* The assignment was confirmed with the following double-resonance combinations:

Signal transition (Ku band)	Pump transition (V band)
5 1 5 - 5 0 5	6 1 5 - 5 0 5
7 1 7 - 7 0 7	8 0 8 - 7 0 7
8 1 8 - 8 0 8	9 1 9 - 8 1 8
	9 0 9 - 8 0 8
9 1 9 - 9 0 9	9 0 9 - 8 0 8
10 1 10 - 10 0 10	10 0 10 - 9 0 9
11 1 11 - 11 0 11	11 1 11 - 10 1 10

** δ_{HFS} is referred to the component with the largest hfs offset from ν_0 .

large deviation for the $J'-J=2-1$ transitions may be due to the fact that K_- reaches the value $K_{-, \text{max}}$ (see Table 4). Transitions with $K_- > 2$ were weighted with 1/100. It is remarkable to notice that the $K_- = 3$ transitions show a positive deviation whereas transitions with $K_- > 3$ are characterized by negative $\Delta\nu$'s. Thus, no obvious functional continuity can be constructed, in contrast to the trans conformer. A centrifugal distortion fit using Van Eijck's sixth order Hamiltonian yields only a gradual improvement.

Figure 5 leads to a slightly modified result. In this case, transitions with $K_- = 2, 3, 4$ were weighted with the factor 1/1000 in a sixth order centrifugal distortion fit (transitions with $K_- > 4$ were fully included in the

fit). The small deviations for transitions with K_- values $K_- < 2$ and $K_- > 4$ suggest that only a few transitions (which means in particular the corresponding energy levels) are perturbed by some effects which have to be discussed hereafter. The frequency data of "cis" cyclopropyl isocyanate are given in Table 4. Table 4 is constructed in the same way as Table 2. ν_0 corresponds

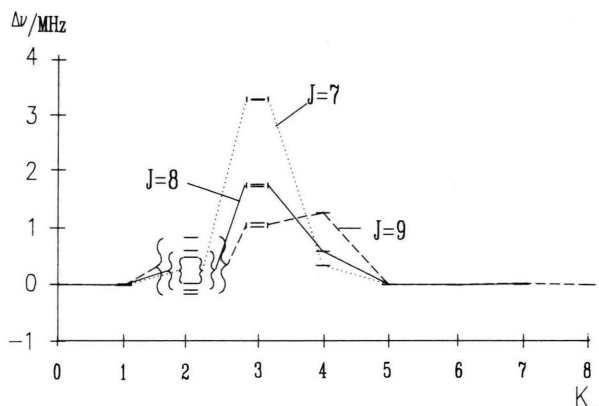


Fig. 5. Deviations $\Delta\nu$ from a sixth order centrifugal distortion fit in dependence on K_- for "cis" cyclopropyl isocyanate. Transitions with $K_- = 2, 3, 4$ are weighted with the factor 1/1000. The braces indicate the related K_- -doublet transitions for each J .

Table 3. Comparison of the rotational and Van Eijck's centrifugal distortion constants of trans cyclopropyl isocyanate for a fourth and sixth order centrifugal distortion fit. The conditions of the fit are given at the bottom of the Table. Numbers in parentheses represent single standard errors in units of the last quoted digit.

	Fourth order	Sixth order
A	16 941.889(8) MHz	16 941.865(3) MHz
B	1 784.3121(6) MHz	1 784.3117(2) MHz
C	1 716.1278(6) MHz	1 716.1281(3) MHz
\tilde{D}_J	0.153(2) kHz	0.159(1) kHz
\tilde{D}_{JK}	52.28(3) kHz	51.72(2) kHz
\tilde{D}_K	-6(3) kHz	-7(1) kHz
$\tilde{\delta}_J$	0.027(1) kHz	0.0176(4) kHz
\tilde{R}_6	-0.019(1) kHz	-0.0179(4) kHz
\tilde{H}_{JK}	-	1.80(7) Hz
\tilde{H}_{KJ}	-	-92.8(3) Hz
Weight of the transitions in the fit:		
	$K_- > 3$: 1/1000	$K_- > 7$: 1/1000
Standard deviation of the fit:		
	14 kHz	5 kHz
Highest correlation:		
	$\tilde{R}_6/\tilde{\delta}_J$: 0.982	$\tilde{R}_6/\tilde{\delta}_J$: 0.854

Table 4. Measured transition frequencies ν_{obs} of “cis” cyclopropyl isocyanate including the quadrupole hyperfine splitting. The hyperfine components are characterized by their overall angular momentum quantum number $F-F'$. δ_{HFS} , $\Delta(\delta_{\text{HFS}})$, \llcorner , n.r., ν_0 : see Table 2. ${}^4\nu_0$: calculated (hypothetical) unsplit line frequencies using Van Eijck’s fourth order centrifugal distortion Hamiltonian. This corresponds to Figure 4. ${}^4\Delta\nu_0 = {}^4\nu_0 - \nu_0$. ${}^6\nu_0$: calculated (hypothetical) unsplit line frequencies using Van Eijck’s sixth order centrifugal distortion Hamiltonian. This corresponds to Figure 5. ${}^6\Delta\nu_0 = {}^6\nu_0 - \nu_0$. In addition, see Table 5 for the conditions of the centrifugal distortion fits. In the table caption it should be: $F-F'$.

J	K ₋	K ₊	J'K ₋ 'K ₊ '	F'-F	ν _{obs}	δ _{HFS}	Δ(δ _{HFS})	ν ₀ 4ν ₀ 6ν ₀	4Δν ₀ 6Δν ₀	J	K ₋	K ₊	J'K ₋ 'K ₊ '	F'-F	ν _{obs}	δ _{HFS}	Δ(δ _{HFS})	ν ₀ 4ν ₀ 6ν ₀	4Δν ₀ 6Δν ₀
					[MHz]	[MHz]	[kHz]	[MHz]	[MHz]						[MHz]	[MHz]	[kHz]	[MHz]	[MHz]
4	0	4	3 0 3	5-4 4-3 3-2	17165.518 ⋈ 17165.595			17165.539 17165.557 17165.538	0.018 -0.001	9	1	8	8 1 7		n. r.			38974.968 38974.958 38974.957	-0.010 -0.011
5	0	5	4 0 4		n. r.			21452.073 21452.061 21452.040	-0.012 -0.033	9	1	9	8 1 8		n. r.			38251.947 38251.940 38251.931	-0.007 -0.016
6	0	6	5 0 5		n. r.			25735.307 25735.327 25735.307	0.020 0.000	4	2	3	3 2 2	5-4 4-3 3-2	17169.993 17170.344 17169.903	-0.351 0.089	-1 -1	17170.087 17170.157 17170.399	0.070 0.312
7	0	7	6 0 6		n. r.			30014.721 30014.737 30014.721	0.016 0.000	5	2	3	4 2 2	6-5 4-3 5-4	21473.667 ⋈ 21473.845			21473.720 21473.771 21473.925	0.051 0.205
8	0	8	7 0 7		n. r.			34289.692 34289.701 34289.691	0.009 -0.001	5	2	4	4 2 3	6-5 4-3 5-4	21461.446 ⋈ 21461.631	-0.185	1	21461.466 21461.553 21461.874	0.087 0.408
9	0	9	8 0 8		n. r.			38559.665 38559.668 38559.669	0.003 0.004	6	2	4	5 2 3	7-6 5-4 6-5	25773.505 ⋈ 25773.609	-0.104	3	25773.538 25773.540 25773.660	0.002 0.122
2	1	1	1 1 0	3-2 2-1 1-0 2-2 1-1	8666.490 8667.282 8665.609 8666.839 8666.753	-0.792 0.881 -0.349 -0.263	0 15 -7 -2	8666.640 8666.554 8666.572	-0.086 -0.068	6	2	5	5 2 4	7-6 5-4 6-5	25752.157 ⋈ 25752.268	-0.111	-2	25752.193 25752.189 25752.601	-0.004 0.408
2	1	2	1 1 1	3-2 2-1 1-0 2-2 1-1	8505.138 8505.939 8504.393 8505.610 8505.188	-0.801 0.745 -0.472 -0.050	1 -1 11 1	8505.296 8505.200 8505.216	-0.096 -0.080	7	2	5	6 2 4	8-7 6-5 7-6	30076.016 ⋈ 30076.085	-0.069	-1	30076.037 30076.004 30076.054	-0.033 0.017
5	1	4	4 1 3		n. r.			21662.737 21662.734 21662.763	-0.004 0.026	8	2	6	7 2 5	9-8 7-6 8-7	34381.588 ⋈ 34381.631	-0.043	-5	34381.600 34381.561 34381.500	-0.039 -0.100
5	1	5	4 1 4		n. r.			21259.673 21259.672 21259.702	-0.001 0.029	8	2	7	7 2 6	9-8 7-6 8-7	34330.606 ⋈ 34330.646	-0.040	9	34330.616 34330.573 34331.216	-0.043 0.600
6	1	5	5 1 4		n. r.			25992.966 25992.963 25992.989	-0.003 0.023	9	2	7	8 2 6		n. r.			38690.518 38690.567 38690.348	0.049 -0.169
6	1	6	5 1 5		n. r.			25509.554 25509.541 25509.571	-0.013 0.017	9	2	8	8 2 7		n. r.			38617.978 38618.021 38618.808	0.043 0.830
7	1	6	6 1 5		n. r.			30321.913 30321.901 30321.921	-0.012 0.008	7	3	4	6 3 3	8-7 6-5 7-6	30047.452 ⋈ 30047.606	-0.154	-6	30047.500 30049.949 30050.771	2.449 3.271
7	1	7	6 1 6		n. r.			29758.317 29758.313 29758.338	-0.004 0.021	7	3	5	6 3 4	8-7 6-5 7-6	30047.084 ⋈ 30047.235	-0.151	-3	30047.131 30049.576 30050.417	2.445 3.286
8	1	7	7 1 6		n. r.			34649.328 34649.315 34649.325	-0.013 -0.003	8	3	5	7 3 4	9-8 7-6 8-7	34342.632 ⋈ 34342.731	-0.099	1	34342.664 34343.501 34344.401	0.837 1.737

Table 4 (continued)

J	K ₋	K ₊	J'K ₋ 'K ₊ '	F'-F	ν_{obs}	δ_{HFS}	$\Delta(\delta_{\text{HFS}})$	ν_0 $4\nu_0$ $6\nu_0$	$4\Delta\nu_0$ $6\Delta\nu_0$
					[MHz]	[MHz]	[kHz]	[MHz]	[MHz]
8	3	6	7 3 5	9-8 7-6 8-7	34341.884 << 34341.984			34341.916 34342.756 34343.693	0.840 1.777
9	3	6	8 3 5	10-9 8-7 9-8	38637.307 << 38637.380			38637.332 38637.392 38638.352	0.060 1.020
9	3	7	8 3 6	10-9 8-7 9-8	38635.940 << 38636.011			38635.964 38636.029 38637.056	0.065 1.092
6	4	2	5 4 1	7-6 6-5 5-4	25753.900 25754.302 25753.815			25754.011 25752.845 25754.227	-1.166 0.216
7	4	3	6 4 2	8-7 7-6 6-5	30046.032 30046.293 30045.995			30046.109 30044.865 30046.452	-1.244 0.343
8	4	4	7 4 3	9-8 8-7 7-6	34337.958 34338.130 34337.939			34338.010 34336.828 34338.608	-1.182 0.598
9	4	5	8 4 4	10-9 8-7 9-8	38629.364 << 38629.496			38629.411 38628.726 38630.687	-0.685 1.276
6	5	1	5 5 0	7-6 6-5 5-4	25751.173 25751.797 25751.018			25751.341 25749.124 25751.350	-2.217 0.009
7	5	2	6 5 1	8-7 7-6 6-5	30042.751 30043.147 30042.674			30042.863 30040.293 30042.863	-2.570 0.000
8	5	3	7 5 2	9-8 8-7 7-6	34334.124 34334.390 34334.085			34334.202 34331.301 34334.202	-2.901 0.000
9	5	4	8 5 3	10-9 9-8 8-7	38625.285 38625.471 38625.265			38625.341 38622.122 38625.339	-3.219 -0.002
7	6	1	6 6 0	8-7 7-6 6-5	30038.955 30039.524 30038.839			30039.113 30035.310 30039.106	-3.803 -0.007
8	6	2	7 6 1	9-8 8-7 7-6	34329.642 34330.030 34329.578			34329.753 34325.448 34329.744	-4.305 -0.009
9	6	3	8 6 2	10-9 9-8 8-7	38620.043 38620.314 38620.007			38620.123 38615.336 38620.116	-4.787 -0.007
8	7	1	7 7 0	9-8 8-7 7-6	34324.763 34325.280 34324.671			34324.909 34318.929 34324.913	-5.980 0.004
9	7	2	8 7 1	10-9 9-8 8-7	38614.433 38614.796 38614.374			38614.537 38607.884 38614.555	-6.653 0.018
9	8	1	8 8 0	10-9 9-8 8-7	38608.359 38608.836 38608.285			38608.497 38599.576 38608.490	-8.921 -0.007

Table 5. The rotational and Van Eijck's centrifugal distortion constants of "cis" cyclopropyl isocyanate for different fit procedures using a fourth (first column) and sixth order (second column) centrifugal distortion Hamiltonian. Furthermore, different sets of transitions are included in the fits (according to Figures 4 and 5). The conditions of the fits are given at the bottom of the Table. Numbers in parentheses represent single standard errors in units of the last quoted digit. n.d.: not determinable, set to zero.

	Fourth order	Sixth order
<i>A</i>	10 215(6) MHz	10 235(3) MHz
<i>B</i>	2 186.859(6) MHz	2 186.856(3) MHz
<i>C</i>	2 106.172(6) MHz	2 106.168(3) MHz
\tilde{D}_J	2.01(1) kHz	2.06(1) kHz
\tilde{D}_{JK}	29.5(3) kHz	21.87(7) kHz
\tilde{D}_K	n.d.	n.d.
$\tilde{\delta}_J$	0.31(2) kHz	0.32(1) kHz
\tilde{R}_6	0.14(1) kHz	n.d.
\tilde{H}_{JK}	—	-1.1(5) Hz
\tilde{H}_{KJ}	—	8(1) Hz
Weight of the transitions in the fit:		
	$K_- > 2: 1/100$	$K_- = 2, 3, 4: 1/1000$
Standard deviation of the fit:		
	53 kHz	26 kHz
Highest correlation:		
	$\tilde{\delta}_J/B: 0.888$	$\tilde{\delta}_J/B: 0.884$

to the fit procedure indicated by Fig. 4, whereas $6\nu_0$ corresponds to Figure 5. Table 5 contains the rotational as well as Van Eijck's centrifugal distortion constants. The "conditions" of the fit are summarized at the bottom of the Table.

Attempt at Explanation

The experimental results of the "cis" isomer indicate that the spectrum of cyclopropyl isocyanate cannot be described completely by application of a centrifugal distortion Hamiltonian. To find an explanation for these effects we have to go back to the basic assumptions of conventional centrifugal distortion theory. In addition to the rigid asymmetric rotor Hamiltonian, centrifugal distortion effects are considered in the total Hamiltonian by a potential energy term [12].

The potential energy is formulated as a harmonic potential function [13]. The elements of the inverse moment of inertia tensor are approximated by a series expansion in the internal displacement coordinates

considering only the linear term. Thus, the theory only regards small displacements of the internal coordinates from their equilibrium values [14]. These conditions are obviously not fulfilled for "cis" cyclopropyl isocyanate. Because of the rigidity of the cyclopropyl frame and of the NCO group especially the torsion around the $C_{\text{frame}}-\text{N}$ bond allows for flexibility of the molecule. If we assume a very flat and anharmonic potential near the "cis" conformation with a low barrier to the trans conformation, the failure of the centrifugal distortion analysis is plausible and the mentioned observation with the beam spectrometer can be explained. This agrees with the statement of Durig et al. who found four excited vibrational states for the trans but only one for the "cis" isomer. They concluded that higher excited states lie above the barrier of internal rotation.

To strengthen the above assumption we tried to calculate the potential function with Gaussian 86 [15]. With a STO-3G basis set a trans and a gauche/gauche minimum potential result by varying the torsion angle in steps of 15° . With an extended 4-31 G* basis set we

obtained a cis and a trans minimum potential when starting the computation of the potential energy function from a cis conformation. This clearly indicates the need of further extended quantum chemical calculations (which we cannot perform presently), and of additional microwave investigations of cyclopropyl isocyanate to obtain a proper understanding of the conformational behaviour and the effects concerning centrifugal distortion.

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